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Nomenclature & Formula Writing 4 Writing Formulas & Names for Polyatomic Compounds
Polyatomic ions, as their name suggests, are ions containing more than one atom. Unlike the atoms combined in binary ionic compounds, the atoms in polyatomic ions are molecules atoms joined by covalent bonds. The nature and properties of covalent bonds are discussed in a later activity, but for now it is im...

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[NAMING IONIC COMPOUNDS WORKSHEET 2](#)

Naming Ionic Compounds Worksheet 2 W 308 Name the following ionic compounds: 1) $\text{Mo}(\text{SO}_4)_3$ 2) $\text{Ti}(\text{NO}_3)_3$ 3) NaH 4) $\text{BeSO}_3 \cdot 7\text{H}_2\text{O}$ 5) $\text{Fe}_3(\text{PO}_4)_2$ 6) $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ 7) $\text{V}(\text{H}_2\text{PO}_4)_4$ 8) $\text{Mn}_3(\text{PO}_4)_7$ 9) $\text{Fe}(\text{CN})_3$ 10) BaSO_4 Write the formulas for the following ionic compounds: 11) vanadium (V) arsenide 12) cobalt (III) nitrite 13) yttrium perchlorate 14) silver phosphide 15) tin (IV) selenide 16) cadmium sulfite 17) nickel (III)...

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[CHEMICAL BONDING](#)

Lab 2: Naming Ionic Compounds Vocabulary cations positively charged ions anions negatively charge ions
cations anions Na^+ Ca^{+2} NH_4^+ Cl^- NO_3^- OH^- Ag^+ Al^{+3} Mg^{+2} SO_4^{2-} PO_4^{3-} ...

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[BINARY IONIC COMPOUNDS WORKSHEET](#)

1. Write the chemical formula for the following binary ionic compounds. a. sodium fluoride h. calcium phosphide b. potassium chloride i. magnesium oxide c. calcium oxide j. aluminum chloride d. lithium bromide k. sodium iodide e. beryllium chloride l. potassium nitride f. potassium sulphide m. lithium sulfide g. magnesium nitride n. barium phosphide 2. Write the name for the following binary ionic...





















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